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The cytotoxicity of brazilin and derivatives might be due to an inhibition of the c-Src-kinase

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Abstract: In this study, several derivatives of brazilin with different lipophilicity were synthesized. The cytotoxic potential of these substances was evaluated in S.R.B. assays. A triacetylated brazilin reaction with PBr₃ or PCl₃ and a subsequent aqueous workup led to the formation of a phosphorous ester containing two triacetylated brazilin subunits. This compound held unexpected high cytotoxicity. In this study, Brazilin-derived triacetate showed good cytotoxic activity (EC50 = 5.2 M) concerning A2780 carcinoma cells. The results from docking studies suggest that brazilin's cytotoxicity might be due to an inhibition of a tyrosine kinase in an ATP-competitive manner.

Keywords: Brazilin; Sappan wood; Cytotoxicity; Tyrosine kinase inhibition.

1. Introduction

Some compounds have been used for many centuries in various applications in different regions of the world. New applications are continually being added, while other uses are almost wholly lost. In this way, some compounds have "survived" for many years. This has undoubtedly also been true for brazilin (1). Brazilin (Fig. 1) is a colorless solid and a precursor for a red dye, brazilein (CI 75280, natural red). It is obtained from tropical hardwoods' heartwood such as Caesalpinia echinata Lamarck, C. sappan L. or Haematoxylum brasiletto Karsten ^{1,2}. In ancient Turkestan, China, Arabia, and India, extract from the wood has been used to dye textiles ³. Especially in Japan, this dying predominantly silk was established in the year 820 CE by the Emperor Saga; the colorant was named "koro". It was used as an official colorant to dye ceremonial costumes. This tradition has survived until today ^{3,4}.

There are also early reports from ancient China of its use as an analgesic ². Furthermore, for these wood chips extracts, antibacterial ^{5,6}, anti-helminthic ² and antioxidant ⁷⁻¹¹ activity has been reported. Recently, its binding mechanism and the inhibitory effect on amyloid aggregation has been investigated by a multi–spectroscopic approach ¹². Furthermore, a neuroprotective effect of 1 has been established in a human neuroblastoma model ¹³. There are also some reports on the cytotoxic activity of brazilein but less for brazilin ^{11,14-21}.

Recently, increased efforts have been made to make **1** the starting point for developing new drugs to be used as anti-acne ², anti-inflammatory ²²⁻²⁵, hypoglycemic, and vasorelaxant agent ²⁶⁻²⁸ or benefiting from its hepatoprotective activity ^{7,30,31}. All these uses would require permanent medication.



Figure 1. Structures of brazilin (1), brazilein, and hematoxylin

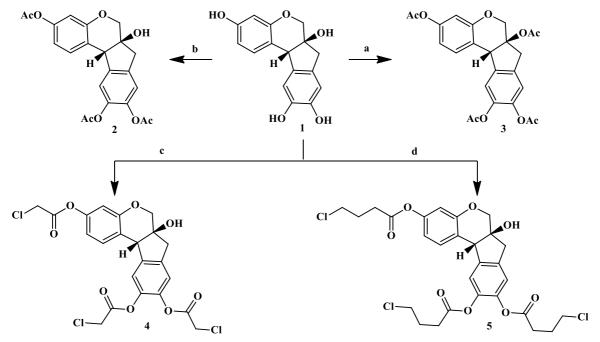
This is in contradiction, however, with those publications reporting minor toxic or cytotoxic effects for **1**. Therefore, we were interested in evaluating the cytotoxic potential of **1**^{18,21}. To also determine a possible influence of lipophilicity on cytotoxic behavior, different acyl derivatives of **1** (log $P_{o/w} = 1.51$) were prepared, such as acetates **2** and **3** (log $P_{o/w} = 2.52$ and 3.00), a tris-chloro-acetate **4** (log $P_{o/w} = 3.45$) and a tris-chlorobutyrate **5** (log $P_{o/w} = 5.11$).

2. Results and Discussion

Sappan wood was obtained from a local supplier. It was shredded and extracted with methanol 32 . The extract was subjected to column chromatography to yield **1** in about 0.7% based on the amount of wood used. This is in good agreement with previously obtained results on the content of **1** in Sappan wood ³. The compound **1** is colorless; however, it oxidizes relatively quickly in the air to red-colored

brazilein ^{33,34}. In some earlier works, this is probably why an off-white or even reddish color of **1** was reported, obviously due to a contamination of **1** with brazilein. Contaminations of the latter compound in samples of **1** can (besides NMR. spectroscopy, M.S., and T.L.C.) also be easily detected by U.V./Vis spectroscopy. The U.V./Vis spectrum of **1** shows the typical p band for aromatics at $\lambda = 289$ nm. On the other hand, another band's presence in the visible range at $\lambda = 445$ nm would suggest that the obtained brazilin is partly contaminated with brazilein.

Acetylation of **1** with acetic anhydride (Scheme 1) yielded the tri-acetate **2** whose acetate groups were detected in the ¹H N.M.R. spectrum ³⁵ at $\delta = 2.20, 2.19$ and 2.18 ppm, respectively. Acetylation with an excess of acetic anhydride furnished the tetra-acetate **3** in 57.8% yield whose acetyl groups were detected at $\delta = 2.26, 2.25, 2.23$ and 2.05 ppm, respectively. A detailed interpretation of I.R. signals of **1** and brazilein has already been reported ³⁶.

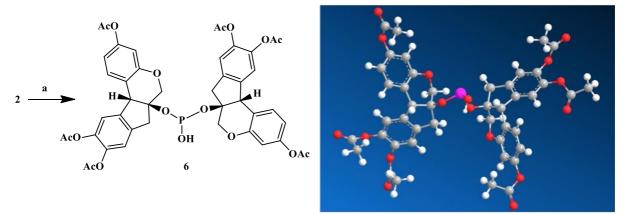


Scheme 1. Reactions and conditions: a. Ac₂O (excess), NEt₃, DMAP (cat.), DCM, 23°C, 12 h, 57.8%; b. Ac₂O (excess), NEt₃, DMAP (cat.), DCM, 23°C, 24 h, 58.4%; c. ClCH₂COCl, NEt₃, DCM, 23°C, 12 h, 66%; d. Cl-(CH₂)₃-COCl, NEt₃, DCM, 23°C, 12 h, 51%

studies showed Previous chloro-acetylated compounds are of higher cytotoxicity than their 37-39 Thus, acetylated analogs 3,9,10-tri-*O*chloroacetyl-brazilin (4) was prepared from 1 and chloroacetyl chloride in 66% isolated yield, while from 1 and 4-chlorobutyryl chloride 5 was obtained in 51% yield. The presence of three chlorine substituents was confirmed not only by the results of the corresponding microanalyses, but also by the corresponding isotope distribution patterns of the [M+Na]⁺ adduct ions detected for 4 and 5 [for 4 m/z 537.0 (100%, ³⁵Cl³⁵Cl), 539.1 (85%, for example ³⁵Cl³⁷Cl³⁵Cl) and 541.0 (28%, for example ³⁷Cl³⁶Cl³⁷Cl were detected)].

The reaction of **2** with PBr₃ or PCl₃ did not result in the formation of a halogenated product. However, after the usual aqueous workup, phosphorous ester **6** (Scheme 2) holding two tri-acetylated brazilin residues was obtained. Its ³¹P N.M.R. chemical shift also characterized this molecule $\delta = 0.53$ ppm as typical for this compound. At room temperature, **6** is a mixture of rotamers leading to a doubling of signals in the respective N.M.R. spectra. On average, the signals differ by about $|\Delta\delta| = 0.06$ ppm except for 1-H and 7-H ($|\Delta\delta| = 0.15$ ppm and $|\Delta\delta| = 0.21$ ppm, respectively). Measurements of N.M.R. spectra at elevated temperatures led to a coalescence of the signals. Six acetyl groups' presence also characterizes **6**, and ESI-MS experiments show a signal at m/z870.7, corresponding to [M+H]⁺. Furthermore, rotations about the central O-P bonds are also hindered due to two bulky residues.

Molecules containing two brazilin subunits are not yet known in the literature. Only a dimeric structure is known, which was created by the reductive coupling of brazilein. Thereby, the coupling of the two subunits took place at position C-4 40 .



Scheme 2. Synthesis of the phosphorous acid derivative 6 from 2 [a) PCl₃, D.C.M., dry pyridine, $0^{\circ}C \rightarrow 23^{\circ}C$, 12 h, 88%], and a ball and stick representation of 6 (main conformer, MMFF94 energy minimized; grey: carbon, white: hydrogen, red: oxygen, pink: phosphorus)

Table 1. Cytotoxicity of compounds **1-6** (EC₅₀ values in μ M from S.R.B. assays after 96 h of treatment, the values are averaged from three independent experiments performed each in triplicate, confidence interval CI = 95%; mean \pm standard mean error, cut-off 30 μ M). Human cancer cell lines: 518A2 (melanoma), A2780 (ovarian carcinoma), HT29 (colorectal adenocarcinoma), MCF-7 (breast adenocarcinoma), A549 (alveolar basal epithelial adenocarcinoma); HeLa (cervical carcinoma), non-malignant: NIH 3T3 (mouse fibroblasts). Staurosporine (**S.T.A.**) was used as a positive standard. Tumor cell/non-tumor cell selectivity S was calculated as S = EC₅₀ [NIH 3T3] / EC₅₀ [cell line].

Compound	518A2	A2780	HT29	MCF-7	A549	HeLa	NIH 3T3
1	9.4 ± 1.2	7.2 ± 1.1	23.1 ± 2.0	17.4 ± 1.9	8.8 ± 1.4	17.0 ± 2.0	12.9 ± 2.4
	(S = 1.4)	(S = 1.8)	(S = 0.6)	(S = 0.7)	(S = 1.4)	(S = 0.7)	
2	8.1 ± 2.5	5.2 ± 1.8	19.6 ± 2.3	18.8 ± 3.5	12.7 ± 2.1	26.4 ± 2.6	15.0 ± 2.3
	(S = 1.9)	(S = 2.9)	(S = 0.8)	(S = 0.8)	(S = 1.2)	(S = 0.6)	
3	14.3 ± 0.8	10.1 ± 1.5	39.4 ± 4.4	23.9 ± 2.4	19.2 ± 0.5	33.4 ± 4.0	15.0 ± 2.3
	(S = 1.9)	(S = 2.7)	(S = 0.7)	(S = 1.1)	(S = 1.4)	(S = 0.8)	
4	7.4 ± 1.0	5.1 ± 0.7	20.5 ± 2.4	14.4 ± 1.7	7.0 ± 0.6	11.9 ± 1.4	2.7 ± 0.6
	(S = 0.4)	(S = 0.5)	(S = 0.1)	(S = 0.2)	(S = 0.4)	(S = 0.2)	
5	5.6 ± 1.7	7.9 ± 1.2	14.9 ± 1.8	12.2 ± 1.0	7.3 ± 1.2	19.4 ± 1.7	9.2 ± 1.3
	(S = 1.6)	(S = 1.2)	(S = 1.7)	(S = 0.8)	(S=1.3)	(S = 0.5)	
6	10.5 ± 3.3	3.9 ± 1.4	18.9 ± 4.3	17.5 ± 3.0	11.7 ± 2.9	20.4 ± 3.0	5.6 ± 1.2
	(S = 0.5)	(S = 1.4)	(S = 0.3)	(S = 0.3)	(S = 0.5)	(S = 0.3)	
STA	0.2 ± 0.02	0.1 ± 0.01	0.1 ± 0.01	0.1 ± 0.05	0.1 ± 0.01	0.1 ± 0.02	0.2 ± 0.02
	(S = 1.0)	(S = 2.0)	(S = 2.0)	(S = 2.0)	(S = 2.0	(S = 2.0)	

To evaluate the compounds' cytotoxicity, sulforhodamine B assays (S.R.B.) were used employing several malignant cell lines and non-malignant fibroblasts (N.I.H. 3T3)^{38,39}. The results from these assays are compiled in Table 1 and depicted in Figure 2.

As a result, the triacetate 2 was – by and large – as cytotoxic as unsubstituted 1. A higher uptake might explain the observed increased activity of 2 due to the greater lipophilicity of 2 compared to 1. Cytotoxicity dropped, however, significantly for the tetra-acetate 3. The tris-chloroacetate 4 was more cytotoxic than the tri-acetate 2 but less selective. This is because hydrolysis releases chloroacetic acid, which

contributes to increased cellular toxicity. However, higher lipophilicity and thus a possibly increased absorption into the cells cannot be excluded. These findings parallel previous results, having been shown for some pentacyclic triterpenoic acid derivatives ⁴¹. The selectivity to discriminate between tumor cells and non-malignant cells was more or less the same for the tris(chlorobutyryl) derivate **5** and unsubstituted **1**. Unexpected high cytotoxicity (EC₅₀ = 3.9 μ M) was observed for **6** and A2780 ovarian carcinoma cells. In general, the highest cytotoxicity of all compounds was observed for 518A2 and A2780 and A549 tumor cells. Mouse fibroblasts were also quite sensitive to these compounds except for compounds **2** and **3**.

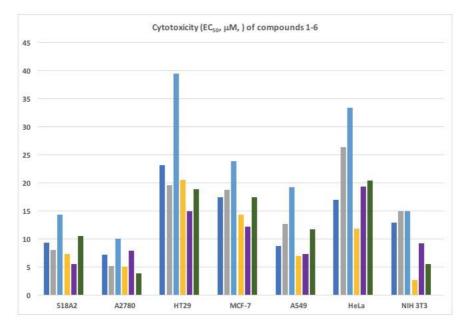


Figure 2. Cytotoxicity (EC₅₀ in μM from SRB assay) of compounds 1–6 [from left to right: 1 (dark blue), 2 (grey), 3 (light blue), 4 (yellow), 5 (pink), 6 (green)]

While the mode of action for the chlorinated acyl derivatives **4** and **5** is likely due to an ability to act as an alkylating agent. A different mechanism might cause the cytotoxicity of 1. Thus, to reveal 1 and its derivatives' activity, a similarity search was conducted employing the SwissTargetPrediction platform (<u>http://www.swisstargetprediction.ch/</u>). The results from this analysis showed that kinases are most likely targeted by **1** and its lipophilic compounds. A 2D similarity search showed that hematoxylin (with an extra hydroxyl group at position 4 compared to **1**, Fig. 1), holds a similarity of 82%. Furthermore, hematoxylin is known as a pan-kinase inhibitor causing cell death.

Lin and colleagues ⁴² could show that especially the c-Src-kinase ⁴³ is inhibited by hematoxylin. Utilizing a docking study based on the crystal structure of

c-Src, they conducted that hematoxylin acts by blocking the active binding site due to an interaction with Glu341 and Met343. We were interested in whether brazilin (1) and its derivatives might act similarly from this intriguing observation. Thus, comparative *in silico* docking experiments were performed. As far as the acylated analogs of brazilin are concerned, we hypothesize that these esters **2-5** might be cleaved ⁴⁴ by intracellular ac(et)ylases.

A comparison between hematoxylin and **1** revealed that both occupy the same binding pocket and interact with the key residues Glu341 and Met343 at the hinge region of c-Src ⁴¹. Additional hydrogen bonds are formed with Ser347 and Asp350 on the other side of the molecule (see Fig. 3). The occupied binding pocket is the adenine pocket of the ATP binding side ⁴⁵.

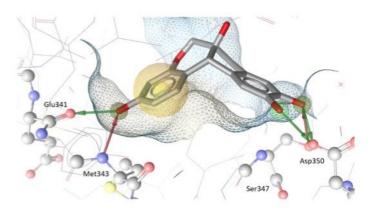


Figure 3. Molecular docking pose of 1 in the kinase domain of c-Src. Hydrogen bonds are shown as red and green arrows, hydrophobic contacts as yellow spheres

Thus, one can assume that **1**, as hematoxylin ⁴², wields its tyrosinase kinase inhibitory activity in an ATPcompetitive manner. The c-Src inhibiting action might explain the detected anticancer activity ⁴⁶ of **1** and its analogs.

3. Conclusion

In this study, brazilin was isolated from sappan wood, and several derivatives of brazilin with different lipophilicity were synthesized, e.g., acetyl derivatives. A reaction of the triacetylated brazilin 2 with PBr₃ or PCl₃ and a subsequent aqueous workup led to the formation of a phosphorous ester containing two triacetylated brazilin subunits. The cytotoxic potential of these substances was evaluated in S.R.B. assays. The phosphorous ester showed unexpected high cytotoxicity (EC₅₀ = 3.9μ M), and the triacetate **2** held good cytotoxicity (EC₅₀ = 5.2 μ M) concerning A2780 carcinoma cells. The results from molecular modeling studies suggest that brazilin's cytotoxic activity might be explained by its ability to inhibit a tyrosine kinase in an ATP-competitive manner. The inhibition of c-Src inhibiting and the enzyme tyrosinase 47 seems thereby of interest and will be investigated in more detail. Future studies will also examine the influence of different substitution patterns and dimeric phosphorous esters' potential.

Acknowledgments

The authors are grateful to Dr. D. Ströhl, Y. Schiller, and S. Ludwig for multiple N.M.R. spectra and the late Dr. R. Kluge for M.S. measurements. U. Lammel performed elemental analyses and optical rotations, and M. Schneider and S. Ludwig recorded U.V./vis and I.R. spectra. Dr. S. Sommerwerk supported some experiments.

4. Experimental

Melting points are uncorrected (Leica hot stage microscope), N.M.R. spectra were recorded using the Varian spectrometers Gemini 2000 or Unity Inova 500 (δ are given in ppm, *J* in Hz; spectra were referenced to residual un-deuterated solvents). The

MS spectra were taken on a Finnigan MAT LCQ (electrospray, spray voltage 4.5 kV, sheath gas nitrogen) instrument. For micro-analyses, a Foss-Heraeus Vario EL (C-HNS) instrument was used. I.R. spectra were measured with a PerkinElmer "Spectrum 1000" FT-IR spectrometer, and U.V./Vis spectra were recorded using a Lambda 750 S U.V./Vis/N.I.R. spectrophotometer from PerkinElmer. For the determination of the specific optical rotation, a P-2000 polarimeter from Jasco was used. T.L.C. was performed on silica gel 60 coated with fluorescent indicator F254 (Merck 5554, detection by U.V. absorption or by treatment with a 10% sulfuric acid solution, ammonium molybdate cerium(IV) sulfate) followed by gentle heating. Starting materials were obtained from local suppliers in bulk, and the solvents were dried according to usual procedures. The $\log P_{o/w}$ values were calculated using the **SwissTargetPrediction** platform (http://www.swisstargetprediction.ch/).

4.1. Cytotoxic evaluation

The cell lines were obtained from the Department of Oncology (Martin-Luther-University Halle Wittenberg). Cultures were maintained as monolayers in RPMI 1640 medium with L-glutamine (Capricorn Scientific GmbH, Ebsdorfergrund, Germany) supplemented with 10% heat-inactivated fetal bovine serum (Sigma-Aldrich Chemie GmbH, Steinheim, Germany) and penicillin/streptomycin (1%, Capricorn Scientific GmbH, Ebsdorfergrund, Germany) at 37°C in a humidified atmosphere with 5% CO₂.

The compounds 'cytotoxicity was evaluated using the sulforhodamine-B (Kiton-Red S, A.B.C.R.) microculture colorimetric assay, as previously reported using confluent cells $^{37-39}$.The EC₅₀ values were averaged from three independent experiments performed each in triplicate calculated from semilogarithmic dose-response curves applying a nonlinear Hills-slope equation (GraphPad Prism5; variables top and bottom were set to 100 and 0, respectively).

4.2. Docking simulation

The molecular docking simulation was carried out with GOLD 5.2 (C.C.D.C., Cambridge, UK). Scoring was calculated using the C.H.E.M.P.L.P. (https://www.ccdc.cam.ac.uk/support-andresources/support/case/?caseid=5d1a2fc0-c93a-49c3-a8e2-f95c472dcff0) scoring function.

PDB entry 1yom ⁴⁵, i.e., the c-Src kinase domain, cocrystalized with the purvalanol B in the ATP binding site, was used as a target structure. The protein was prepared in GOLD, using default settings. The binding site was defined on the A chain in an 8 Å radius at the co-crystalized ligand site. The Protein-ligand interactions of the docking poses were analyzed using LigandScout 4.2(www.inteligand.com/ligandscout). A re-docking of purvalanol B was conducted to optimize the docking settings. The best-ranked pose with the final docking settings had an R.M.S.D. of 0.711 compared to the cocrystallized ligand.

Brazilin [7,11b-Dihydro-(6aS,11bR) benz [b]indeno[1,2-d]pyran-3,6a,9,10(6H)-tetrol] (1)

Sappan wood (500 g; Kremer Pigmente GmbH, Aichstetten, Germany) was shredded and extracted with methanol (3 x 5 L, each for 1 day). The solvent was distilled off, and the residue subjected to column chromatography (SiO₂, CHCl₃/MeOH, 10:1) to yield **1** (3.3 g, 0.7%) as a colorless solid; m.p. = 157–161°C (lit.:⁴⁸ 156–157°C); $[\alpha]_D^{20} = +84^\circ$ (*c* 0.32, MeOH), [lit.:⁴⁹ $[\alpha]_D = +90^\circ$ (*c* 0.10, MeOH)]; R_F = 0.55 (CHCl₃/MeOH, 4:1);

IR (KBr): v = 3422br, 1624m, 1508m, 1464m, 1458m, 1384ss, 1324m, 1302m, 1234m, 1158m, 1116m, 1072w, 1036m cm⁻¹;

UV/Vis (MeOH): λ_{max} (log ε) = 289 (3.76) nm;

¹H NMR (500 MHz, CD₃OD): δ = 7.18 (dd, J = 8.3, 0.7 Hz, 1H, 1-H), 6.71 (d, J = 0.9 Hz, 1H, 11-H), 6.60 (s, 1H, 8-H), 6.47 (dd, J = 8.3, 2.5 Hz, 1H, 2-H), 6.29 (d, J = 2.5 Hz, 1H, 4-H), 3.96 (s, 1H, 11b-H), 3.92 (dd, J = 11.3, 1.4 Hz, 1H, 6-H_a), 3.69 (d, J = 11.3 Hz, 1H, 6-H_b), 3.02 (d, J = 15.6 Hz, 1H, 7-H_a), 2.67 (d, J = 15.7 Hz, 1H, 7-H_b) ppm;

¹³C NMR (125 MHz, CD₃OD): δ = 157.9 (C-3), 155.7 (C-4a), 145.6 (C-9), 145.3 (C-10), 137.5 (C-11a), 132.2 (C-1), 131.3 (C-7a), 115.6 (C-1a), 112.9 (C-8), 112.4 (C-11), 109.9 (C-2), 104.3 (C-4), 78.1 (C-6a), 70.9 (C-6), 51.1 (C-11b), 42.9 (C-7) ppm;

MS (ESI, MeOH): *m*/*z* 285.2 (66%, [M-H]⁻), 570.9 (100%, [2M-H]⁻);

analysis calcd for $C_{16}H_{14}O_5$ (286.28): C 67.13, H 4.93; found: C 66.87, H 5.11.

3,9,10-Tri-O-acetyl-brazilin (2)

Brazilin (1, 508.9 mg, 1.8 mmol) was suspended in dry DCM (30 mL), triethylamine (0.27 mL, 2.0 mmol), acetic anhydride (535.0 mg, 5.2 mmol) and DMAP (cat.) were added, and the mixture was stirred at 23°C for 12 h. Usual aqueous work-up followed by column chromatography (SiO₂, hexane/ethyl acetate, 1:1) gave **2** (422 mg, 58.4%) as a colorless solid; m.p.

148–152°C (lit.: 112°C ⁴⁰; 105-106 °C ⁵⁰); $[\alpha]_D^{20} = +65^\circ$ (*c* 0.38, CHCl₃); $R_F = 0.30$ (hexane, ethyl acetate, 1:1);

IR (KBr): v = 3446br, 1768s, 1616w, 1590w, 1498m, 1488m, 1430m, 1372m, 1214vs, 1168m, 1148s, 1122m, 1070w, 1040m, 1014m cm⁻¹; UV/Vis (MeOH): λ_{max} (log ε) = 278 (3.72) nm;

¹H NMR (500 MHz, CDCl₃): δ = 7.25 (dd, J = 8.4, 0.6 Hz, 1H, 1-H), 7.02 (d, J = 1.1 Hz, 1H, 11-H), 6.94 (s, 1H, 8-H), 6.69 (dd, J = 8.3, 2.4 Hz, 1H, 2-H, 6.58 (d, J = 2.2 Hz, 1H, 4-H), 4.06 (s, 1H, 11b-H), 3.91 (dd, J = 11.6, 1.5 Hz, 1H, 6-H_a), 3.7 (d, J = 11.5 Hz, 1H, 6-H_b), 3.09 (d, J = 16.4 Hz, 1H, 7-H_a), 2.83 (d, J = 16.6 Hz, 1H, 7-H_b), 2.20 (s, 3H, CH₃), 2.19 (s, 3H, CH₃), 2.18 (s, 3H, CH₃) ppm;

¹³C NMR (125 MHz, CDCl₃): δ = 169.33 (C-12), 168.43 (C-16), 168.33 (C-14), 154.0 (C-4a), 150.0 (C-3), 142.3 (C-11a), 141.22 (C-9), 140.82 (C-10), 137.7 (C-7a), 130.8 (C-1), 120.0 (C-8), 119.2 (C-11), 119.0 (C-1a), 114.7 (C-2), 110.4 (C-4), 76.9 (C-6a), 69.5 (C-6), 50.1 (C-11b), 41.2 (C-7), 20.83 (CH₃), 20.43 (CH₃), 20.33 (CH₃) ppm;

MS (ESI, MeOH): *m/z* 430.1 (100%, [M+NH₄]⁺), 435.2 (19%, [M+Na]⁺), 444.9 (22%, [M+H+MeOH]⁺), 846.7 (14%, [2M+Na]⁺);

analysis calcd for $C_{22}H_{20}O_8$ (412.39): C 64.08, H 4.89; found: C 63.85, H 5.03.

3,6a,9,10-Tetra-O-acetyl-brazilin (3)

To a suspension of brazilin (1, 500.0 mg, 1.8 mmol) in dry D.C.M. (35 mL), D.M.A.P. (cat.), triethylamine (1.25 mL, 9.0 mmol) and acetic anhydride (0.76 mL, 8.0 mmol) were added, and the mixture was stirred at 23°C for 12 h. Usual aqueous workup followed by a recrystallization from acetone/water gave **3** (460 mg, 57.8%) as a colorless solid; m.p. 150–151°C (lit.:⁵⁰ 149–151°C; 147.0-148.4 ⁵¹); $[\alpha]_D^{20} = +64^\circ$ (*c* 0.35, CHCl₃) (lit.:⁵² $[\alpha]_D^{20} = +76.4^\circ$ (HOAc)) ; $R_F = 0.64$ (hexane/ethyl acetate, 1:1);

IR (KBr): v = 3440br, 1770s, 1742s, 1618w, 1592w, 1500m, 1490m, 1432m, 1370s, 1324w, 1312w, 1210vs, 1150s, 1106m, 1070m, 1046w, 1014m cm⁻¹; UV/Vis (MeOH): λ_{max} (log ε) = 274 (3.66),

277 (3.67) nm;

¹H NMR (500 MHz, DMSO-d₆): δ = 7.59 (dd, J = 8.3, 0.7 Hz, 1H, 1-H), 7.20 (s, 1H, 8-H), 7.17 (d, J = 1.2 Hz, 1H, 11-H), 6.83 (dd, J = 8.3, 2.4 Hz, 1H, 2-H), 6.70 (d, J = 2.4 Hz, 1H, 4-H), 4.65 (dd, J = 12.4, 1.8 Hz, 1H, 6-H_a), 4.59 (s, 1H, 11b-H), 3.68 (d, J = 12.4 Hz, 1H, 6-H_b), 3.49 (d, J = 16.9 Hz, 1H, 7-H), 3.39 (d, J = 17.2 Hz, 1H, 7-H), 2.25 (s, 3H, CH₃), 2.24 (s, 3H, CH₃), 2.23 (s, 3H, CH₃), 2.05 (s, 3H, CH₃) ppm;

¹³C NMR (125 MHz, DMSO-d₆): δ = 170.23 (C-18), 169.13 (C-12), 168.33 (C-16), 168.23 (C-14), 153.6 (C-4a), 150.0 (C-3), 141.52 (C-11a), 141.12 (C-9), 140.22 (C-10), 136.8 (C-7a), 131.3 (C-1), 120.4 (C-8), 119.0 (C-11), 118.3 (C-1a), 150.3 (C-2), 110.3 (C-4), 83.1 (C-6a), 65.3 (C-6), 47.7 (C-11b), 38.8 (C-7), 21.53 (CH₃), 20.83 (CH₃), 20.33 (CH₃), 20.32 (CH₃) ppm;

MS (ESI, MeOH): *m*/*z* 472.1 (60%, [M+NH₄]⁺), 477.1

(100%, [M+Na]⁺);

analysis calcd for $C_{24}H_{22}O_9$ (454.43): C 63.43, H 4.88; found: C 63.11, H 5.01.

3,9,10-Tris-O-chloroacetyl-brazilin (4)

To a suspension of brazilin (**1**, 400 mg, 1.4 mmol) in dry D.C.M. (15 mL), triethylamine (1.0 mL, 7.2 mmol) was added; the mixture was cooled to 0°C, and chloroacetyl chloride (725 mg, 6.4 mmol) in dry D.C.M. (15 mL) was slowly added, and stirring at 23°C was continued for 12 h. Usual aqueous workup followed by column chromatography (SiO₂, hexane/ethyl acetate, 1:1) gave **4** (480 mg, 66%) as an off-white oil; $[\alpha]_D^{20} = +16^\circ$ (*c* 4.15, CHCl₃); $R_F = 0.46$ (hexane/ethyl acetate, 1:1);

IR (KBr): v = 3442br, 2956w, 1778s, 1616m, 1594w, 1498m, 1486m, 1430m, 1408m, 1310s, 1282m, 1242s, 1154vs, 1126vs, 1040m cm⁻¹;

UV/Vis (MeOH): λ_{max} (log ϵ) = 278 (3.79) nm;

¹H NMR (500 MHz, CDCl₃): δ = 7.33 (dd, J = 8.4, 0.6 Hz, 1H, 1-H), 7.12 (d, J = 1.1 Hz, 1H, 11-H), 7.10 (s, 1H, 8-H), 6.81 (dd, J = 8.3, 2.4 Hz, 1H, 2-H), 6.70 (d, J = 2.4 Hz, 1H, 4-H), 4.28 (s, 2H, CH₂ (13)), 4.25– 4.24 (m, 4H, 15-H₂ + 17-H₂), 4.18 (s, 1H, 11b-H), 4.03 (dd, J = 11.6, 1.6 Hz, 1H, 6-H_a), 3.80 (d, J = 11.6 Hz, 1H, 6-H_b), 3.26 (d, J = 16.4 Hz, 1H, 7-H_a), 2.95 (d, J = 16.6 Hz, 1H, 7-H_b) ppm;

¹³C NMR (125 MHz, $CDCl_3$): $\delta = 166.03$ (C-12), 165.23 (C-16), 165.13 (C-14), 154.3 (C-4a), 150.0 (C-3), 142.9 (C-11a), 140.82 (C-9), 140.52 (C-10), 138.4 (C-7a), 131.3 (C-1), 120.2 (C-8), 119.4 (C-1a), 119.3 (C-11), 114.9 (C-2), 110.6 (C-4), 77.4 (C-6a), 69.9 (C-6), 50.4 (C-11b), 41.2 (C-7), 41.03 (C-13), 40.53 (C-15), 40.53 (C-17) ppm;

MS (ESI, MeOH): *m*/*z* 537.0 (100%, [M+Na]⁺), 539.1 (85%, [M+Na]⁺), 541.0 (28%, [M+Na]⁺);

analysis calcd for $C_{22}H_{17}O_8Cl_3$ (515.72): C 51.24, H 3.32; found: C 50.97, H 3.05.

3,9,10-Tris-O-(4-chlorobutyryl)-brazilin (5)

Following the procedure for the synthesis of **4**, from **1** (401 mg, 1.4 mmol) and 4-chlorobutyrylchloride (880 mg, 6.4 mmol) followed by column chromatography (hexane/ethyl acetate, 1:1) **5** (430 mg, 51%) was obtained as an off-white oil; $[\alpha]_D^{20} = +29^\circ$ (*c* 0.51, CHCl₃); $R_F = 0.51$ (hexane/ethyl acetate, 1:1);

IR (KBr): v = 3446m, 2966w, 2926w, 2870w, 1762vs, 1616m, 1592m, 1498m, 1488m, 1440m, 1430m, 1382m, 1370m, 1310s, 1268m, 1216s, 1150vs, 1126vs, 1040s cm⁻¹;

UV/Vis (MeOH): λ_{max} (log ϵ) = 278 (3.78) nm;

¹H NMR (500 MHz, CDCl₃): δ = 7.34 (dd, J = 8.4, 0.6 Hz, 1H, 1-H), 7.06 (d, J = 1.2 Hz, 1H, 11-H), 7.04 (s, 1H, 8-H), 6.79 (dd, J = 8.3, 2.4 Hz, 1H, 2-H), 6.68 (d, J = 2.4 Hz, 1H, 4-H), 4.18 (s, 1H, 11b-H), 4.03 (dd, J = 11.5, 1.6 Hz, 1H, 6-H_a), 3.82 (d, J = 11.5 Hz, 1H, 6-H_b), 3.67–3.64 (m, 6H, 15-H₂), 19-H₂, 23-H₂), 3.28 (d, J = 16.3 Hz, 1H, 7-H_a)), 2.93 (d, J = 16.4 Hz, 1H, 7-H_b), 2.76–2.71 (m, 6H, 13-H₂), 17-H₂, 21-H₂), 2.20–2.15 (m, 6H, 14-H₂, 18-H₂, 22-H₂) ppm;

¹³C NMR (125 MHz, CDCl₃): δ = 171.23 (C-12),

170.33 (C-16), 170.23 (C-20), 154.1 (C-4a), 150.1 (C-3), 142.2 (C-11a), 141.32 (C-9), 140.92 (C-10), 137.5 (C-7a), 131.1 (C-1), 120.2 (C-8), 119.3 (C-11), 118.9 (C-1a), 115.2 (C-2), 110.7 (C-4), 77.3 (C-6a), 69.9 (C-6), 50.4 (C-11b), 43.93 (C-15), 43.83 (C-19), 43.83 (C-23), 41.0 (C-7), 31.33 (C-13), 30.83 (C-17), 30.73 (C-21), 27.53 (C-14), 27.43 (C-18), 27.33 (C-22) ppm; MS (ESI, MeOH): m/z 616.1 (100%, [M+NH₄]⁺), 621.1 (78%, [M+Na]⁺), 623.1 (64%, [M+Na]⁺), 625.1 (20%, [M+Na]⁺);

analysis calcd for $C_{28}H_{29}O_8Cl_3$ (599.88): C 55.06, H 4.87; found: C 54.84, H 5.00.

(Hydroxyphosphanediyl)bis[oxy(6a*S*,11b*R*)-7,11 b-dihydrobenzo[*b*]indeno[1,2-*d*]pyran-6a, 3,9,10 (6*H*)-tetrayl] hexaacetate (6)

To a suspension of **2** (51.3 mg, 0.12 mmol) in dry DCM (3 mL), dry pyridine (0.05 mL) was added. The mixture was cooled, and at 0°C, PCl₃ (0.02 mL, 0.21 mmol) was added. Stirring at 23°C was continued for 12 h. Usual aqueous work-up followed by an extraction with diethyl ether (3 x 20 mL) and column chromatography (SiO₂, hexane/ethyl acetate, 1:2) gave **6** (62 mg, 88%) as a colorless solid; m.p. 149-151°C; $[\alpha]_D^{20} = +9^\circ$ (*c* 4.4, CHCl₃); $R_F = 0.54$ (hexane/ethyl acetate, 1:2);

IR (KBr): v = 3442m, 2930w, 1770vs, 1618m, 1592m, 1500s, 1490m, 1430m, 1372s, 1308m, 1272m, 1210vs, 1170s, 1150s, 1132s, 1070m, 1046m, 1012s, 970s cm⁻¹;

UV/Vis (MeOH): $λ_{max}$ (log ε) = 278 (3.87) nm;

¹H NMR (400 MHz, CDCl3): $\delta = 7.29$ (d, J = 8.4 Hz, 1H, 1-H), 7.14 (d, J = 8.4 Hz, 1H, 1'-H), 7.03 (d, J = 1.1 Hz, 1H, 11-H), 7.02–7.01 (m, 3H, 8-H, 8'-H), 11'-H), 6.80 (dd, J = 8.3, 2.3 Hz, 1H, 2-H), 6.75 (dd, J = 8.3, 2.3 Hz, 1H, 2'-H), 6.66 (d, J = 2.3 Hz, 1H, 4-H), 6.59 (d, J = 2.3 Hz, 1H, 4'-H), 4.55–4.42 (m, 4H, 6-H_a + 6'-H_a + 11_b-H) + 11b'H), 3.88–3.75 (m, 3H, 6-H_b + 6'-H_b) + 7-Ha), 3.54 (d, J = 16.3 Hz, 1H, 7'-H_a), 3.16 (d, J = 9.3 Hz, 1H, 7-H_b), 3.11 (d, J = 9.1 Hz, 1H, 7'-H_b), 2.28–2.24 (m, 18H, 6 x CH₃) ppm;

¹³C NMR (100 MHz, CDCl₃): δ = 169.33 (C-12 + C-12'), 168.43 (C-16), 168.33 (C-16'), 168.33 (C-14 + C-14'), 153.9 (C-4a + C-4a'), 150.7 (C-3), 150.5 (C-3'), 141.9 (C-11a), 141.9 (C-11a'), 141.62 (C-9), 141.52 (C-9'), 140.82 (C-10), 140.82 (C-10'), 136.2 (C-7a), 136.1 (C-7a'), 130.8 (C-1), 130.8 (C-1'), 120.4 (C-8), 120.2 (C-8'), 119.5 (C-11), 119.5 (C-11'), 117.8 (C-1a), 117.6 (C-1a'), 115.5 (C-2 + C2'), 110.8 (C-4), 110.7 (C-4'), 85.8 (C-6a), 85.7 (C-6a'), 85.6 (C-6), 85.5 (C-6'), 49.5 (C11-b), 49.4 (C-11b'), 40.6 (C-7), 40.4 (C-7'), 21.22 (CH₃, C-13 + C-13'), 20.73 (CH₃, C-15), 20.73 (CH₃, C-15'), 20.73 (CH₃, C-17), 20.73 (CH₃, C-17') ppm;

³¹P NMR (162 MHz, CDCl₃): δ = 0.53 (s, P-OH) ppm; MS (ESI, MeOH): *m/z* 870.7 (18%, [M+H]⁺), 887.9 (35%, [M+NH₄]⁺), 893.0 (100%, [M+Na]⁺), 1763.5 (15%, [2M+Na]⁺);

analysis calcd for $C_{44}H_{39}PO_{17}$ (870.74): C 60.69 H 4.51; found: C 60.41, H 4.67.

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