

Synthesis of new spiroheterocycles-fused isoxazoline from 2-arylidenes-3-phenyl-1-indanones through a regio- and diastereospecific 1,3-dipolar cycloaddition

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Abstract: New spiroisoxazolines **3** have been synthesized by 1,3-dipolar cycloaddition of aryl nitrile oxides with 2-arylidenes-3-phenyl-1-indanones. The reaction occurs in a regio- and diastereospecific manner and leads to one cycloadduct in all the cases. The proposed structure of the obtained cycloadducts was established based on spectroscopic data and confirmed by radiocrystallographic study. The spectral data were in favor of the observed regiochemistry and diastereoselectivity of this reaction.

Keywords: Spiroisoxazoline; 2-arylidenes-3-phenyl-1-indanones; 1,3-dipolar cycloaddition; regioselectivity; stereoselectivity.

1. Introduction

Organic compound-fused 2-isoxazoline frameworks are a very interesting heterocycle in both organic and medicinal chemistry ¹⁻⁸. Isoxazolines also serve as essential building blocks for synthesizing a wide range of biologically active compounds ^{9,10}. For these reasons, a great deal of work has been devoted to synthesizing these kinds of heterocycles ^{11,12}.

In addition, the spiroisoxazoline scaffold forms an integral part of many beneficial biologically active compounds. Indeed, spiroisoxazolines exhibit an extensive array of biological and pharmacological activities such as antiviral ¹³, anticancer ^{14,17,18}, antimalarial ¹⁵, antituberculosis ¹⁶, antibacterial ^{17,19} and antiparasitic ²⁰.

Spiroisoxazolines synthesized in our laboratory in 2005 have been patented for their antituberculosis and anti-HIV properties ^{21,22}. Since then, we have been further developing these highly promising molecules ^{16,20}.

In the current work, we used 2-arylidenes-3-phenyl-1-indanones to synthesize spiroisoxazolines by reacting with aryl nitrile oxide. The introduction of a phenyl group at the 3-position of the (E)-2-

(4-arylidene)-indanones made the reaction of 1,3-dipolar cycloaddition with aryl nitrile oxides not only regio- and diastereospecific, but also stereospecific, and only the cycloadduct "anti" was obtained, making this a more efficient method.

2. Results and Discussion

The 1,3-dipolar cycloaddition reaction of alkynes/alkenes with nitrile oxides, respectively, constitutes a versatile tool to access isoxazole and isoxazoline heterocycles ²³⁻²⁵. In this context, and in continuation of our ongoing research devoted to synthesis heterocyclic compounds with promising biological and pharmacological activities via 1,3-dipolar cycloaddition across dipolarophiles with exocyclic double bond with several 1,3-dipoles ^{16,20}. We describe a reaction between aryl nitrile oxide precursors (arylaloximes) and arylidene-indanone derivatives in one pot, to give the desired spiroisoxazolines **3** (Scheme 1). We carried out the reaction with 2-arylidenes-3-phenyl-1-indanones and a slight excess of aryl nitrile oxide precursors in chloroform at a temperature not exceeding 5°C (Scheme 1). Aryl nitrile oxides were generated

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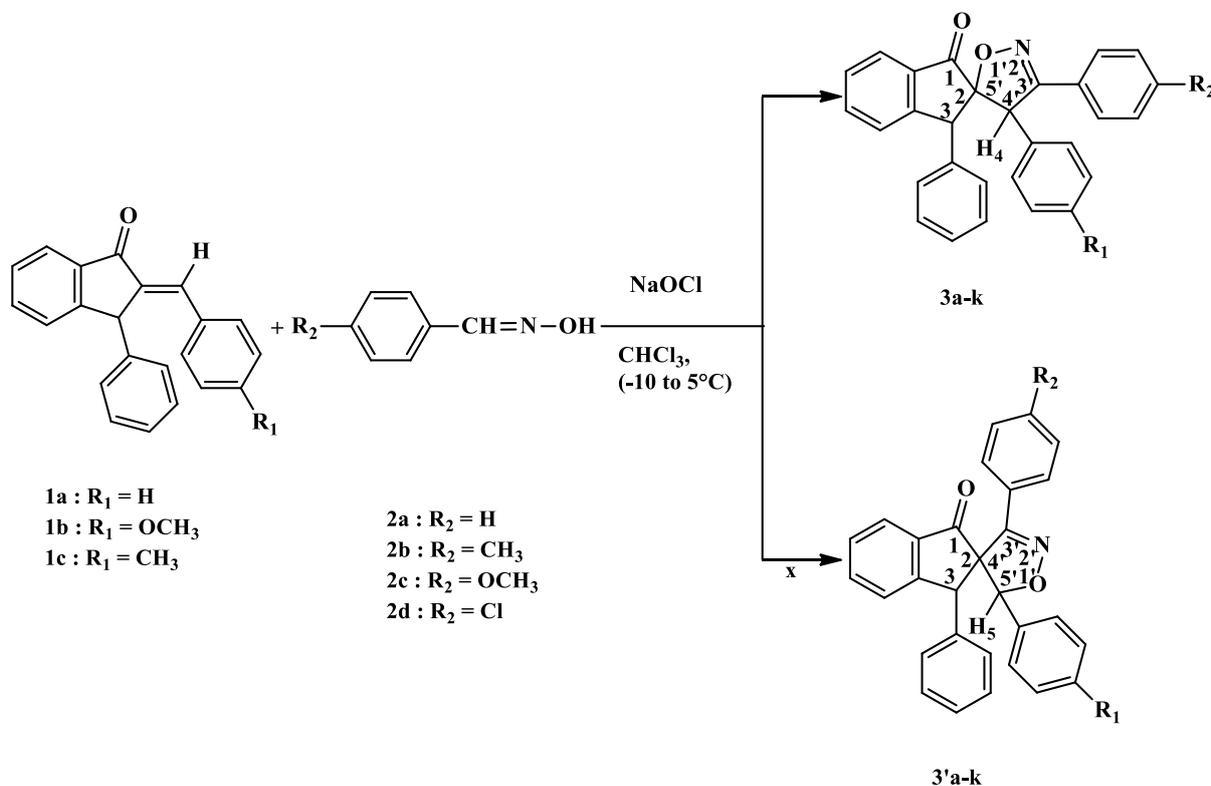
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"in situ" from corresponding arylaldoximes by action of sodium hypochlorite at low temperature so there is no dimerization of the dipole to furoxane. There are many ways to generate aryl nitrile oxide. The most commonly used ones are dehydration of nitro compounds²⁶ or the action of triethylamine on hydroxamoyl chlorides²⁷. We chose to generate our aryl nitrile oxide "in situ" by the action sodium

hypochlorite on arylaldoximes because of its simplicity²⁸. The 2-arylidene-3-phenyl-1-indanones used as starting materials in this work were obtained through a Knoevenagel condensation of appropriate aldehyde on 3-phenyl-1-indanone in the presence of a basic medium, according to previously published protocol^{29,30}.



Scheme 1. Synthesis of spiroisoxazolines **3** (regiospecificity of the reaction)

NMR spectra of the crude product revealed the formation of only the regioisomer 3-phenyl-3',4'-diaryl-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-ones **3** (Scheme 1). The regiochemistry shown in this reaction was similar to the one obtained in reactions using other dipolarophiles³¹⁻³⁷.

The structural assignment of the isolated cycloadducts **3** was made based on spectroscopic data (IR, NMR), which are summarized in the experimental part. The isolated cycloadducts were considered to be regioisomers of type **3**, not regioisomers of type **3'**. This was based on the chemical shifts of the isoxazoline ring protons (H₄) compared to analogous cycloadducts. Thus the isoxazoline ring protons of **3** resonate at 4.4 ppm. In contrast, the isoxazoline ring protons H₅ of cycloadducts **3'** would be expected at higher chemical shift values due to the effect of oxygen atom compared with analogous cycloadducts³⁸.

Moreover, the ¹³C NMR spectrum reveals two characteristic signals at 199 ppm and 53 ppm,

corresponding to C=O carbon and benzylic carbon C₄. Furthermore, the signal of spiranic carbon (C_{2,5'}) appears at 96 ppm, which agrees with the literature for the proposed regiochemistry³⁹. If we suppose that **3'** compounds are obtained, the spiranic carbon (C_{2,4'}) would have appeared at a lower shift since it would be far from the oxygen atom. These spectral data are in favor of regioisomer **3** and confirm the regiospecificity of the action of aryl nitrile oxides on 2-arylidene-3-phenyl-1-indanones. In addition, the FT-IR spectrum of the cycloadduct **3** shows two significant bands around 1700 cm⁻¹ and 1600 cm⁻¹ assigned, respectively, to C=O and C=N stretching. We also noticed in most cases that the yield of the reaction is higher when dipole **2a** (R=H) is used. This can be explained by the less steric hindrance.

In the ¹H NMR spectrum of cycloadduct **3b** (Figure 1), the signal of the methyl appears as a singlet at 2.28 ppm, while the signals of H₄ and H₃ appear at singlets at 4.42 and 4.75 ppm, respectively. The signals of the aromatic protons appear as multiplet between 6.36 and 7.95 ppm.

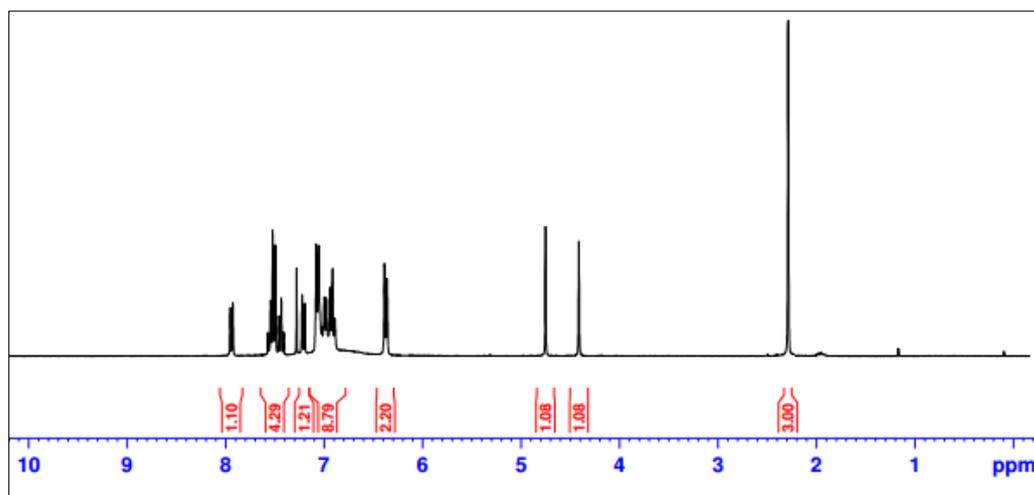
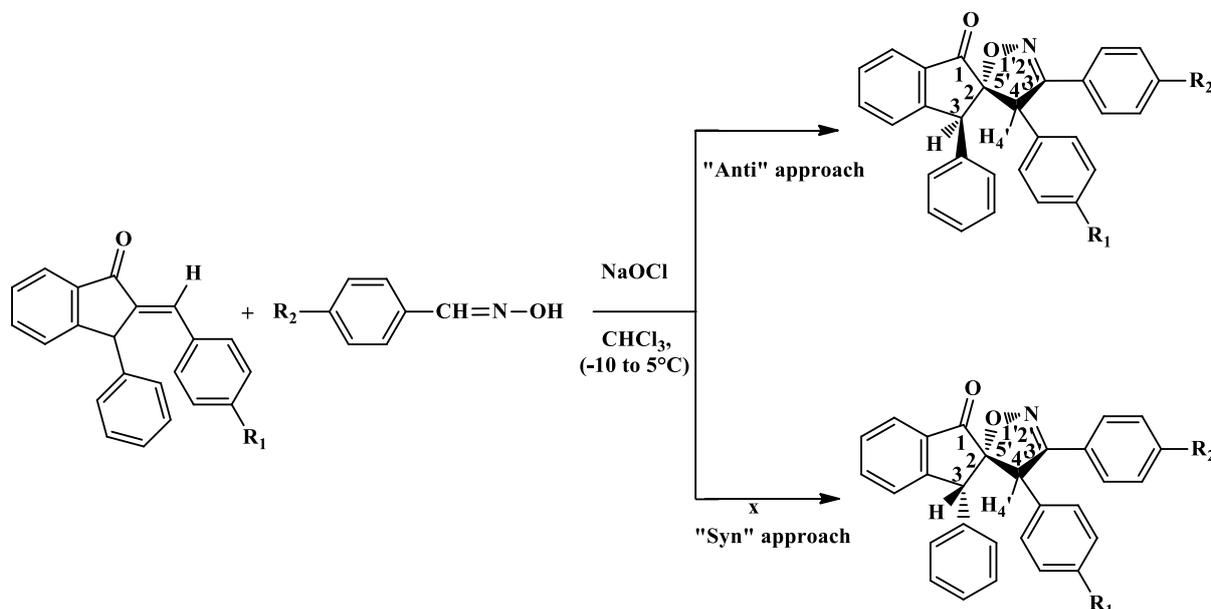


Figure 1. ^1H NMR spectrum of **3b** (CDCl_3)

The stereoselectivity of 1,3-dipolar cycloaddition reactions is well known. This reaction is accompanied by the creation of one or two centers of chirality on the dipolarophile. If both faces of the latter are equivalent, the dipole can react from both sides of the dipolarophile with an equal probability, and then a racemic mixture is obtained. Diastereoselectivity is possible if both sides of the dipolarophile are diastereotopic. Each of the two possible approaches

(syn and anti-approaches) would lead to a different diastereoisomer since the reaction studied here is regioselective. Obtaining a single cycloadduct in all cases, while both approaches are theoretically possible, makes it possible to affirm that the reaction is diastereospecific. The dipole approach toward the dipolarophile is always from the opposite side of the phenyl substituent of the dipolarophile (Scheme 2).



Scheme 2. Diastereospecificity of the reaction

The structure of cycloadduct **3b**, whose crystallographic data have been previously published³¹, was corroborated by X-ray single-crystal

diffraction (Figure 2). This structural determination confirms the resulting regioselectivity and diastereospecificity.

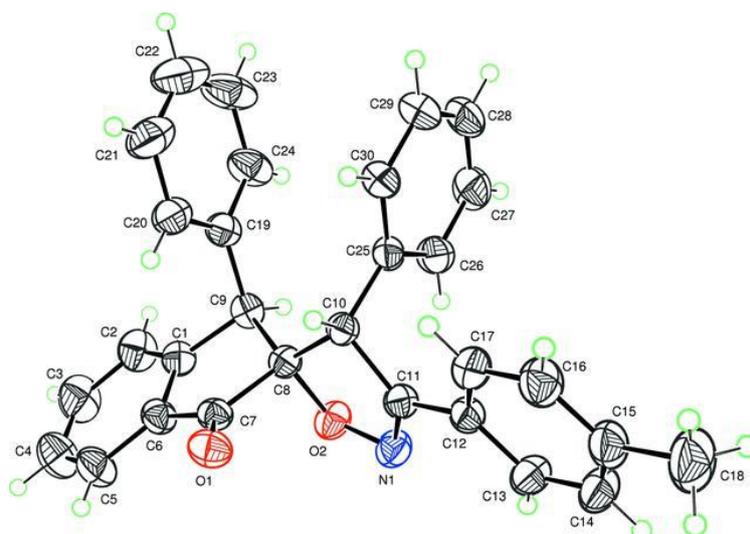


Figure 2. ORTEP diagram of cycloadduct **3b***

*The molecular structure of the title compound with the atom labeling scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are represented as small circles.

3. Conclusion

In summary, we have shown in this work that the spiro-isoxazolines **3** can easily be prepared by 1,3-dipolar cycloaddition of 2-arylidene-3-phenyl-1-indanones **1** with the appropriate aryl nitroxide. We have also shown that these reactions were regiospecific and diastereospecific, due to the influence of steric factors, which are important in the stereoselectivity of the dipole-dipolarophilic approach. The high diastereospecificity and regiospecificity were determined based on spectroscopic data and corroborated by radiocrystallographic study.

4. Acknowledgements

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5. Experimental

5.1. General

Chemicals and reagents were purchased from commercial sources. All melting points were determined using a Stuart SMP 10 melting point apparatus. The infrared spectra were recorded at room temperature using BRUKER VERTEX 70 spectrometer. The ^1H and ^{13}C NMR spectra were recorded at room temperature on BRUKER AVANCE II 300 MHz instrument. Aryldoximes and 2-arylidene-3-phenyl-1-indanones were prepared using previously reported procedures^{38,40}.

5.2. General procedure of the preparation of spiro-isoxazolines: **3**

In a 100 ml flask, (2E)-2-benzylidene-3-phenyl-2,3-dihydro-1H-inden-1-one (2 mmol) and appropriate

aryldoxime (2.4 mmol) were dissolved in chloroform (20 ml). The mixture was stirred at 263 K in an ice-brine bath, then 15 ml of sodium hypochlorite (NaOCl) were added dropwise, keeping the mixture below 278 K. The mixture was left under magnetic stirring for 16 hours at room temperature until completion of the reaction. The mixture was washed with water until neutral pH and dried over anhydrous sodium sulfate. The solvent was evaporated under vacuum, and the obtained crude oil was crystallized in ethanol to obtain the spirocycloadducts **3**.

5.3. Characterization of compounds

3,3',4'-triphenyl-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (**3a**)

White solid: yield = 85 %; Mp. = 296°C.

^1H NMR (300 MHz, CDCl_3) (δ /ppm): 4.41(s, H_4); 4.76(s, H_3); 6.35-7.95(m, H, aromatic).

IR (KBr) cm^{-1} : $\nu(\text{C}=\text{O}) = 1711$; $\nu(\text{C}=\text{N}) = 1596$.

^{13}C NMR (300 MHz, CDCl_3) (δ /ppm): 51.85 (C_3); 53.61 (C_4); 96.08 ($\text{C}_{2,5'}$); 125.19; 126.25; 126.40; 127.12; 127.28; 127.56; 128.46; 128.60; 130.13; 132.81; 134.59; 135.96; 140.22; 155.33; 162.20; 199.26 ($\text{C}=\text{O}$).

3-4'-diphenyl-3'-(p-tolyl)-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (**3b**)

White solid: yield = 70%; Mp. = 266°C.

IR (KBr) cm^{-1} : $\nu(\text{C}=\text{O}) = 1712$; $\nu(\text{C}=\text{N}) = 1597$.

^1H NMR (300MHz, CDCl_3) (δ /ppm): 2.28(s, 3H); 4.41(s, H_4); 4.75(s, H_3); 6.36-7.95(m, H, aromatic).

^{13}C NMR (300 MHz, CDCl_3) (δ /ppm): 21.34(C_{CH_3}); 51.93 (C_3); 53.74 (C_4); 95.91($\text{C}_{2,5'}$); 125.12; 125.65; 126.20; 126.36; 127.03; 127.31; 127.49; 128.38; 128.45; 128.52; 129.19; 132.93; 134.77; 135.82; 140.31; 155.33; 162.16; 199.22($\text{C}=\text{O}$).

3'-(4-methoxyphenyl)-3,4'-diphenyl-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (3c)

White solid; yield = 75%; Mp. = 258°C.

IR (KBr) cm⁻¹: $\nu(\text{C=O}) = 1712$; $\nu(\text{C=N}) = 1597$.**¹H NMR (300MHz, CDCl₃) (δ /ppm):** 3.75(s, 3H); 4.40(s, H_{4'}); 4.73(s, H₃); 6.36-7.95(m, H. aromatic).**¹³C NMR (300 MHz, CDCl₃) (δ /ppm):** 51.93(C₃); 53.84(C_{4'}); 55.21(C_(OCH₃)); 95.82(C_{2,5'}); 113.98; 121.00; 125.12; 126.20; 126.36; 127.03; 127.32; 128.37; 128.46; 128.54; 129.10; 132.95; 134.84; 135.81; 140.33; 155.34; 161.05; 161.80; 199.29(C=O).**3'-(4-chlorophenyl)-3,4'-diphenyl-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (3d)**

White solid; yield = 80%; Mp. = 228°C.

IR (KBr) cm⁻¹: $\nu(\text{C=O}) = 1712$; $\nu(\text{C=N}) = 1594$.**¹H NMR (300MHz, CDCl₃) (δ /ppm):** 4.42(s, H_{4'}); 4.72(s, H₃); 6.31-7.96 (m, H. aromatic)**¹³C NMR (300 MHz, CDCl₃) (δ /ppm):** 51.85(C₃); 53.58 (C_{4'}); 96.25(C_{2,5'}); 125.17; 126.30; 126.39; 127.01; 127.28; 128.51; 128.68; 128.75; 128.81; 132.77; 134.31; 135.99; 136.12; 140.12; 155.30; 161.27; 199.07(C=O).**4'-(4-methoxyphenyl)-3,3'-diphenyl-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (3e)**

White solid; yield = 70%; Mp. = 260°C.

IR (KBr) cm⁻¹: $\nu(\text{C=O}) = 1715$; $\nu(\text{C=N}) = 1603$.**¹H NMR (300MHz, CDCl₃) (δ /ppm):** 3.74 (s, 3H); 4.39 (s, H_{4'}); 4.72 (s, H₃); 6.40-7.55 (m, H. aromatic)**¹³C NMR (300 MHz, CDCl₃) (δ /ppm):** 51.90 (C₃); 52.89 (C_{4'}); 55.26 (C_(OCH₃)); 95.98 (C_{2,5'}); 114.08 ;125.15; 126.33; 126.41; 126.64 ;127.28; 127.57; 128.44; 128.49; 130.08; 132.92; 135.92; 140.43; 155.33; 158.73; 162.25; 199.48(C=O).**4'-(4-methoxyphenyl)-3-phenyl-3'-(p-tolyl)-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (3f)**

White solid; yield = 80%; Mp. = 292°C.

IR (KBr) cm⁻¹: $\nu(\text{C=O}) = 1713$; $\nu(\text{C=N}) = 1603$.**¹H NMR (300MHz, CDCl₃) (δ /ppm):** 2.29(s, 3H); 3.73(s, 3H); 4.38(s, H_{4'}); 4.71(s, H₃); 6.39-7.56 (m, H. aromatic)**¹³C NMR (300 MHz, CDCl₃) (δ /ppm):** 21.41(C_(CH₃)); 51.92(C₃); 52.93(C_{4'}); 55.25(C_(OCH₃)); 95.81(C_{2,5'}); 114.04; 125.13; 125.63; 126.30; 126.40; 126.80; 127.29; 127.51; 128.39; 128.43; 129.22; 132.97; 135.87; 140.33; 140.48; 155.34; 158.67; 162.21; 199.48(C=O).**3',4'-bis(4-methoxyphenyl)-3'-phenyl-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (3g)**

White solid; yield = 79%; Mp. = 279°C.

IR (KBr) cm⁻¹: $\nu(\text{C=O}) = 1710$; $\nu(\text{C=N}) = 1605$.**¹H NMR (300MHz, CDCl₃) (δ /ppm):** 3.74(s, 3H); 3.76(s, 3H); 4.38(s, H_{4'}); 4.68(s, H₃); 6.39-7.94 (m, H. aromatic)**¹³C NMR (300 MHz, CDCl₃) (δ /ppm):** 51.92(C₃); 53.05(C_{4'}); 55.26(C_(OCH₃)); 95.72(C_{2,5'}); 120.97; 125.4.12; 126.30; 126.40; 126.87; 127.29; 128.39; 128.42; 129.12; 132.98; 135.86;

140.49; 155.34; 158.67; 160.98; 161.85; 199.49(C=O).

3'-(4-chlorophenyl)-4'-(4-methoxyphenyl)-3-phenyl-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (3h)

White solid; yield = 76%; Mp. = 281°C.

IR (KBr) cm⁻¹: $\nu(\text{C=O}) = 1714$; $\nu(\text{C=N}) = 1595$.**¹H NMR (300MHz, CDCl₃) (δ /ppm):** 3.74(s, 3H); 4.39(s, H_{4'}); 4.68(s, H₃); 6.38-7.95 (m, H. aromatic)**¹³C NMR (300 MHz, CDCl₃) (δ /ppm):** 51.85(C₃); 52.80(C_{4'}); 55.27(C_(OCH₃)); 95.82(C_{2,5'}); 113.97; 114.17; 126.29; 127.01; 127.27; 129.74; 132.82; 133.45; 136.03; 136.07; 140.30; 155.31; 158.84; 161.31; 199.30(C=O).**3,3'-diphenyl-4'-(p-tolyl)-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (3i)**

White solid; yield = 88%; Mp. = 264°C.

IR (KBr) cm⁻¹: $\nu(\text{C=O}) = 1713$; $\nu(\text{C=N}) = 1604$.**¹H NMR (300MHz, CDCl₃) (δ /ppm):** 2.24(s, 3H); 4.40(s, H_{4'}); 4.73(s, H₃); 6.35-7.57 (m, H. aromatic)**¹³C NMR (300 MHz, CDCl₃) (δ /ppm):** 21.07(C_(CH₃)); 51.92(C₃); 53.26(C_{4'}); 95.81(C_{2,5'}); 125.15; 126.24; 126.41; 127.33; 127.57; 128.31; 128.42; 128.49; 129.25; 130.07; 131.50; 135.92; 136.78; 140.31; 155.58; 157.97; 201.27(C=O).**3-phenyl-3',4'-di-p-tolyl-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (3j)**

White solid; yield = 69%; Mp. = 190°C.

IR (KBr) cm⁻¹: $\nu(\text{C=O}) = 1712$; $\nu(\text{C=N}) = 1602$.**¹H NMR (300MHz, CDCl₃) (δ /ppm):** 2.30(s, 3H); 2.30(s, 3H); 4.39(s, H_{4'}); 4.71(s, H₃); 6.35-7.96 (m, H. aromatic).**¹³C NMR (300 MHz, CDCl₃) (δ /ppm):** 21.43(C_(CH₃)); 21.43(C_(CH₃)); 51.96(C₃); 53.34(C_{4'}); 95.82(C_{2,5'}); 124.21; 125.11; 125.89; 126.19; 126.39; 126.92; 127.50; 127.59; 127.84; 128.28; 128.37; 128.97; 129.21; 131.55; 134.90; 135.74; 135.83; 194.69(C=O).**3'-(4-methoxyphenyl)-3-phenyl-4'-(p-tolyl)-4'H-spiro[indene-2,5'-isoxazol]-1(3H)-one (3k)**

White solid; yield = 65%; Mp. = 208°C.

IR (KBr) cm⁻¹: $\nu(\text{C=O}) = 1712$; $\nu(\text{C=N}) = 1597$.**¹H NMR (300MHz, CDCl₃) (δ /ppm):** 2.29(s, 3H); 3.76(s, 3H); 4.38(s, H_{4'}); 4.69(s, H₃); 6.35-7.94 (m, H. aromatic).**¹³C NMR (300 MHz, CDCl₃) (δ /ppm):** 21.05(C_(CH₃)); 51.95(C₃); 53.44(C_{4'}); 55.23(C_(OCH₃)); 95.82(C_{2,5'}); 114.04; 125.13; 125.63; 126.30; 126.40; 126.80; 127.29; 127.51; 128.39; 128.43; 129.22; 132.97; 135.87; 140.33; 140.48; 155.34; 158.67; 162.21; 194.69(C=O).**X-Ray crystallography for 3b: (3,4'-Diphenyl-3'-p-tolyl-4'H-spiro[indan-2,5'-[1,2]oxazol]-1-one)**In the title compound, C₃OH₂₃NO₂, the five-membered rings are both in envelope conformations with the same spiro C atom as the flap. The benzene ring and the two phenyl rings are inclined to the mean

plane of the indene ring system by 83.98 (8), 81.46 (8), and 72.31 (7) °. In the crystal, molecules are linked by pairs of C-H...O hydrogen bonds into inversion dimers. The dimers are further connected by C-H...N interactions, forming layers' parallel to (101).

Data collection

Bruker X8 APEX diffractometer.

38803 measured reflections, 5942 independent reflections, 3783 reflections with $I > 2\sigma(I)$, $R_{int} = 0.042$

Crystal data

$M_r = 429.49$, Monoclinic, $P2_1 = n$, $a = 9.7381$ (7) Å, $b = 20.5072$ (14) Å, $c = 11.8261$ (8) Å, $\beta = 102.836$ (2)°, $V = 2302.7$ (3) Å³, $Z = 4$, Mo $K\alpha$ radiation, $\mu = 0.08$ mm⁻¹, $T = 296$ K, $0.42 \times 0.31 \times 0.26$ mm.

Further crystallographic data can be obtained free of charge from the Cambridge Crystallographic Data Center through www.ccdc.cam.ac.uk/data_request/cif.CCDC1431561

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